

Incorporating Detailed Water Chemistry into Process-Scale Cost Optimization with Machine Learning

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Chemistry governs operation of water treatment processes, but is difficult to model

WaterTAP typically uses property packages for non-electrolyte solutions

- Non-electrolyte -> components are water and salt (i.e., NaCl, TDS, etc)
 - Properties =

VAVVI

f(*salt concentration, temperature, pressure*)

- Good for bulk properties like density, osmotic pressure, viscosity, specific enthalpy
- 2. Electrolyte -> components are water and all the potential species
 - Must track numerous electrolyte species and chemical reactions
 - Essential for ion activities, solubility/scaling tendencies, precipitation



Components	Carbonation Process		
Na	H2O ←→ H+ + OH-		
К	CaOH+		
Ca	CaHCO3+ ←→ Ca2+ + HCO3-		
Mg	CaCO3 $\leftarrow \rightarrow$ Ca2+ + CO32-		
CI	H2CO3 $\leftarrow \rightarrow$ CO2 (aq) + H2O		
HCO3	H2CO3 ←→ H+ + HCO3-		
Si	HCO3- $\leftarrow \rightarrow$ H+ + CO32- 2		

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Са

Mg

CI

SO4

HCO3

Si

- Electrolyte theoretical models have numerous terms and parameters to represent all of the interactions (e.g., MSE, Pitzer, eNRTL)
- Data availability limits the species that can be considered
 - Inherently large models with many complications
 - Numerous species and reactions
- 2. Electroly
- Species can be at 0 concentration and increase by many orders of magnitude (round-off errors can be problematic)
- Must track numerous electrolyte species and chemical reactions
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nation Process H+ + OH- $\leftarrow \rightarrow$ Ca2+ + OH-CaHCO3+ \leftarrow \rightarrow Ca2+ + HCO3- $CaCO3 \leftrightarrow Ca2+ + CO32 H2CO3 \leftrightarrow CO2 (aq) + H2O$ $H2CO3 \leftrightarrow H+ + HCO3-$ HCO3- $\leftarrow \rightarrow$ H+ + CO32-3



WaterTAP has 3 approaches for water chemistry

All approaches use external water chemistry software:

- 1. Narrow surrogate models
 - Inputs are the key decision variables of the flowsheet
 - Polynomial functions, Radial basis functions (interpolative model), etc.
- 2. Broad surrogate models
 - Inputs are apparent species concentrations, pH, pressure, temperature
 - Neural net (machine learning model)
- 3. Direct integration (Demo on Thursday at 5 PM)
 - Use pyomo <u>External Grey Box Model</u>
 - Requires the external water chemistry software to provide the Jacobian (and Hessian)
 - Possible with <u>Reaktoro</u>-pse (Repository on watertap-org)





Building and integrating surrogate models

- 1. Generate relevant brine scenarios
- 2. Use OLI Cloud API to calculate properties for brine scenarios
- 3. Use IDAES tools to fit or integrate models into IDAES compatible models
- 4. Use WaterTAP flowsheet with the OLI surrogate model



Our first approach was narrow surrogate models

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Broad surrogate models are needed to assess different train configurations and feed compositions

- Large number of treatment train configurations
 - Use of recycle loops ullet
 - Multiple stages
 - Different combinations of driving forces
- Water compositions vary dramatically across US

Machine learning models can enable generation of broad surrogate models

- Enables adding neural networks to Pyomo models:
- Supports dense and convolution layers
- Supports a number of activation functions
- Supports Keras and ONNX standards

Key Questions:

(1) Can deep neural networks provide broad range of chemistry estimates?

(2) How does NN architecture impact solver like IPOPT

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Tailoring data sampling is key for good accuracy

Data size and NN architecture play a "secondary" role in accuracy

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Large networks and tanh are key for use in EO frameworks

NN accuracy tested against USGS brackish water data set (non-synthetic data) Solvability tested using NNs in a black box desalination model using USGS brackish water data set Solved using IPOPT with MA27 linear solver – tested 500 different feed compositions and 2 different guesses

NNs enable assessment of complex treatment trains

PHREEQC ML models for:

- pH control
- precipitation
- mineral scaling

Flowsheets contained about 30 NN models with 30,000 parameters each Solved in <5 min.

18 decision variables (degrees of freedom):

- 3 HCl acid doses, 3 antiscalant doses
- 2 lime doses, 2 soda ash doses
- 3 RO design and operating variables
- 2 MVC design and operating variables

	lon (mg/L)	Case 1	Case 2
	Na	739	1120
	CI	870	1750
	K	9	15
	Ca	258	150
	Mg	90	33
	SO4	1011	260
	HCO3	385	250
	Sr	3	0.08
	SiO2	25	30.5
I S	TDS	3397	3609

Are our networks also suggesting we add "glue" to fix water treatment?

NN accuracy for Scaling tendency *prediction*:

• Average error: 0.9%

NAWI

- 95th percentile of error: ~5%
- 99th percentile of error: 18.5%

Out of 56 simulations, 6 points had poor estimates

NNs provide great accuracy on "average" but can unpredictably and rapidly degrade in performance.

Reaktoro-pse enables exact calculation, enabling us for the first time to "fact check" the ML surrogates

Reaktoro-PSE integrates Reaktoro chemistry models directly into IDEAS and IDAES compatible libraries.

Reaktoro-PSE blocks are applied to estimate track changes in:

- pH
- Scaling tendencies
- Precipitation amount
 Uses same database as ML
 models imitating them as closely as
 possible

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There is no "best" method, but Reaktoro-pse is a great starting point

	Narrow surrogates Broad ML surrogates		Reaktoro-pse	
	PySmo a black-boyed rodd ling tool	MLT	Reaktoro for Python and C++	
Data quantity need	100-100,000 pts	500,000-1,000,000 pts	N/A	
Data tailoring	None to high	High	N/A	
Training time	10-600 seconds	>600 seconds	N/A	
Computational intensity	Very low (1-2x increase)	Low to Mid (2-5x increase)	Mid to high (5-50x increase)	
Stability in IPOPT	Medium (local minimum issues)	Medium (local minimum issues)	TBD (~preliminary stability is high, but sensitive to model and Jacobian scaling)	
Error in estimates	~0-10% - depends on surrogates	~0-30% Depending on breadth of model and components Suffers from edge case errors	Exact solution	

Thank you

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Thank you

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