

Optimal Schedule, Design, and Operation of Solid Oxide Electrolysis Cell Systems Accounting for Long-Term Performance and Health Degradation

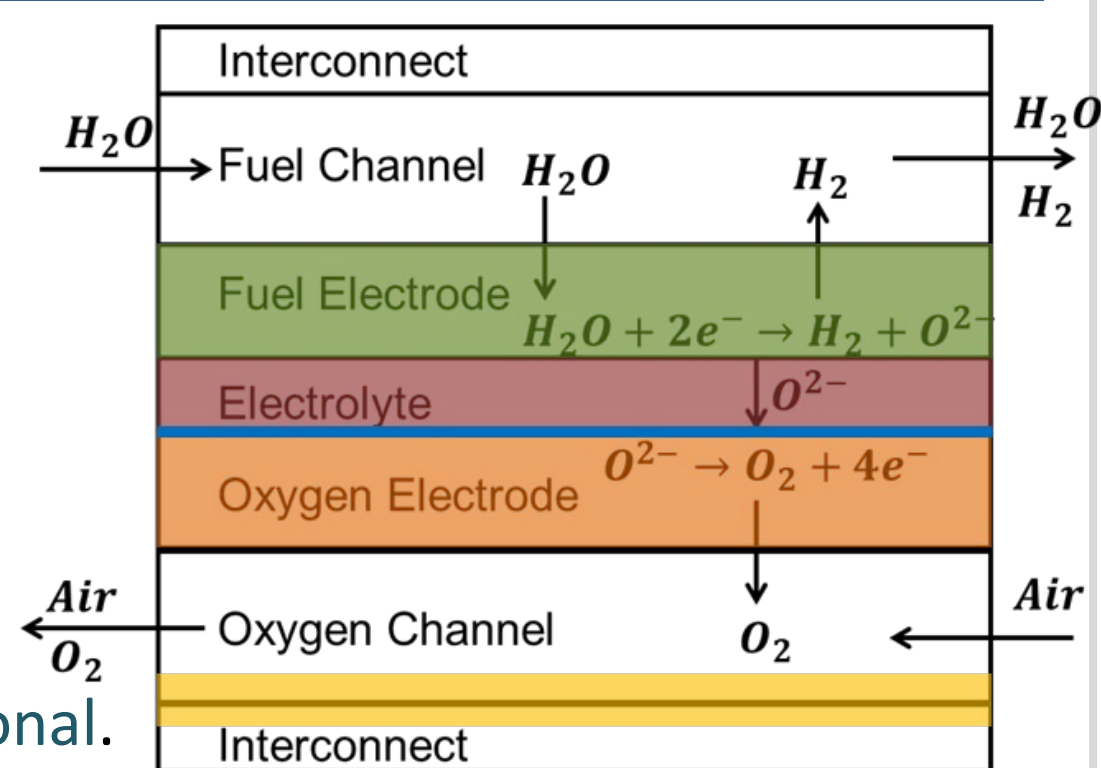
Nishant V. Giridhar^a, Debangsu Bhattacharyya^a, Douglas A. Allan^{b,c}, Mingrui Li^d, Lorenz T. Biegler^d, Stephen E. Zitney^c, Eric Liese^c
^aWest Virginia University, ^bNETL Support Contractor, ^cNational Energy Technology Laboratory, ^dCarnegie Mellon University

Background

Solid-Oxide Cells (SOC) produce H₂ through steam electrolysis at high theoretical efficiency and low direct emissions.

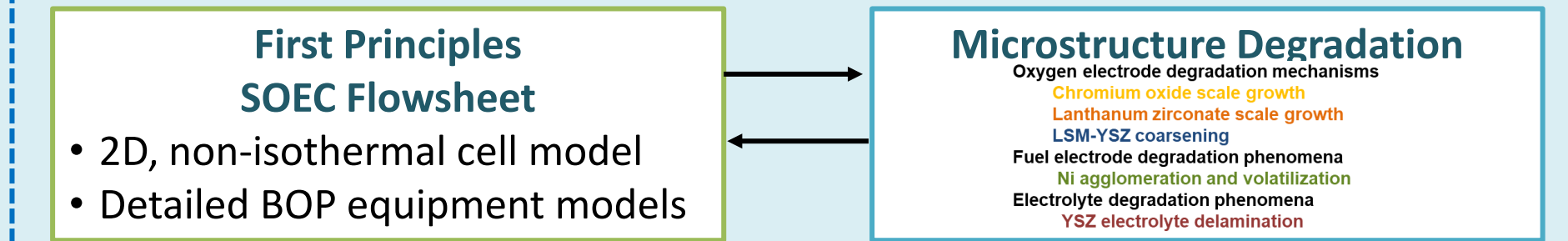
Slow microstructure degradation phenomena decrease efficiency and can require premature replacement of the SOC.

Reducing degradation is often considered a materials design problem, rather than operational.



Combating degradation through long-term optimization

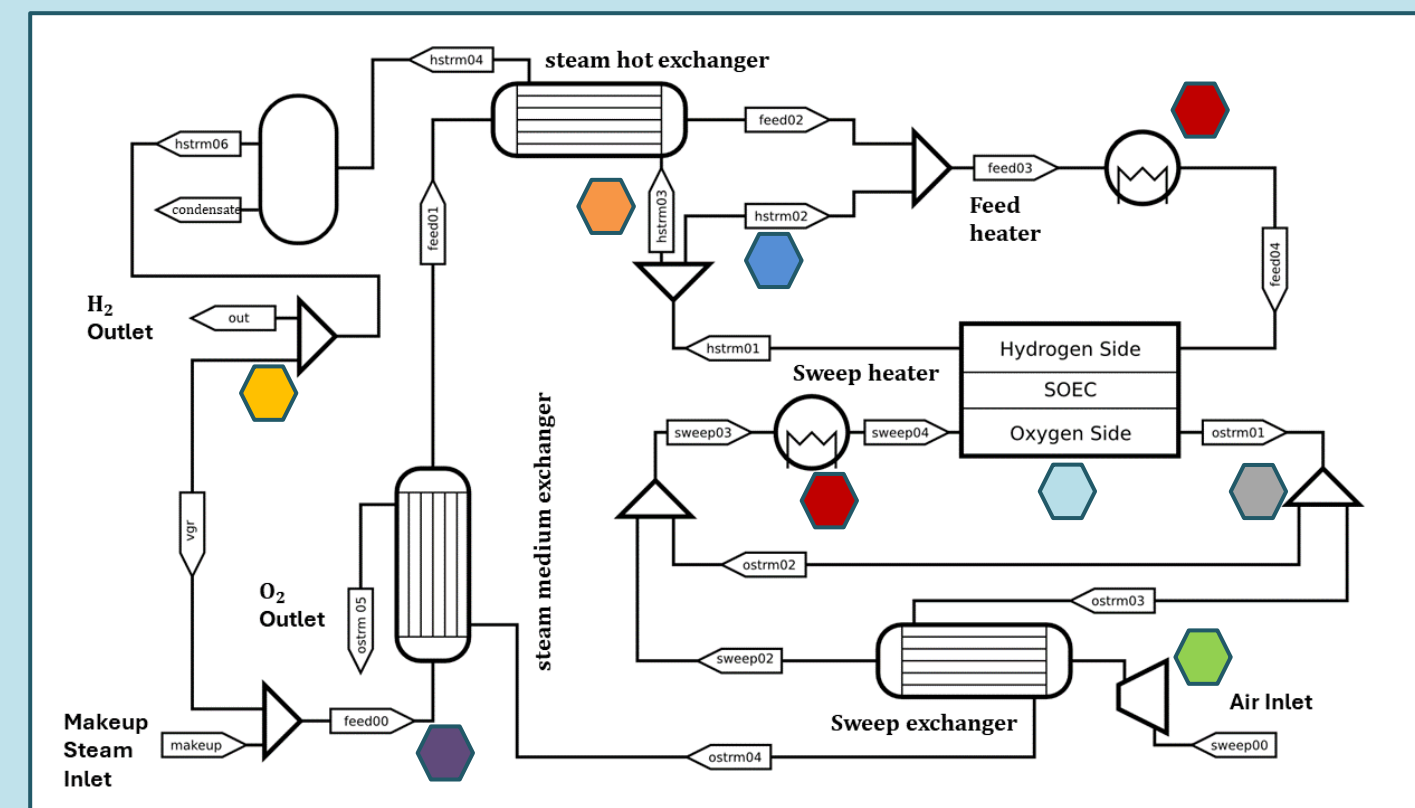
Optimization
minimize $f(x)$



Operating Decisions

1. Flowsheet and cell level setpoints

Symbol	Decision Variable
Cell Potential / Current Density	Cell Potential / Current Density
Feed recycle splitter outlet H ₂ O mole fraction	Feed recycle splitter outlet H ₂ O mole fraction
Feed/ Sweep electric heater duties	Feed/ Sweep electric heater duties
Condenser splitter recycle split fraction	Condenser splitter recycle split fraction
Feed medium exchanger inlet flowrate	Feed medium exchanger inlet flowrate
Feed recycle splitter split fraction	Feed recycle splitter split fraction
Sweep recycle splitter split fraction	Sweep recycle splitter split fraction
Sweep blower molar flowrate	Sweep blower molar flowrate



Cell degradation impacts the operation of the Balance of Plant (BOP) and decreases system efficiency.

2. Choice of long-term operating mode

Potentiostatic
(Constant Voltage)

Galvanostatic
(Constant H₂ Production)

Flexible

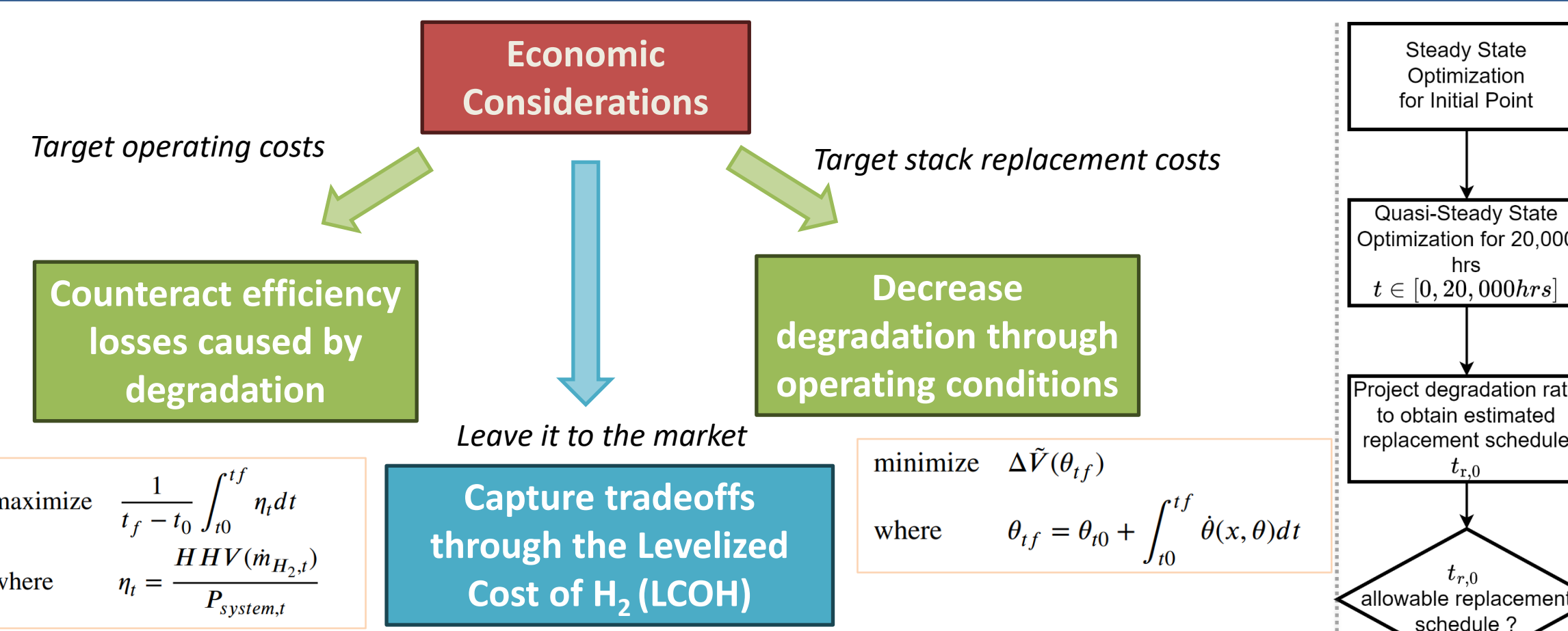
A flexible mode of operation allows both potential and current density to vary simultaneously.

3. Selection of replacement schedule

$$CAPEX = CC_{BOP} + n_{stack} CC_{stack}$$

Efficiency loss due to degradation can be counteracted by more frequent stack replacements.

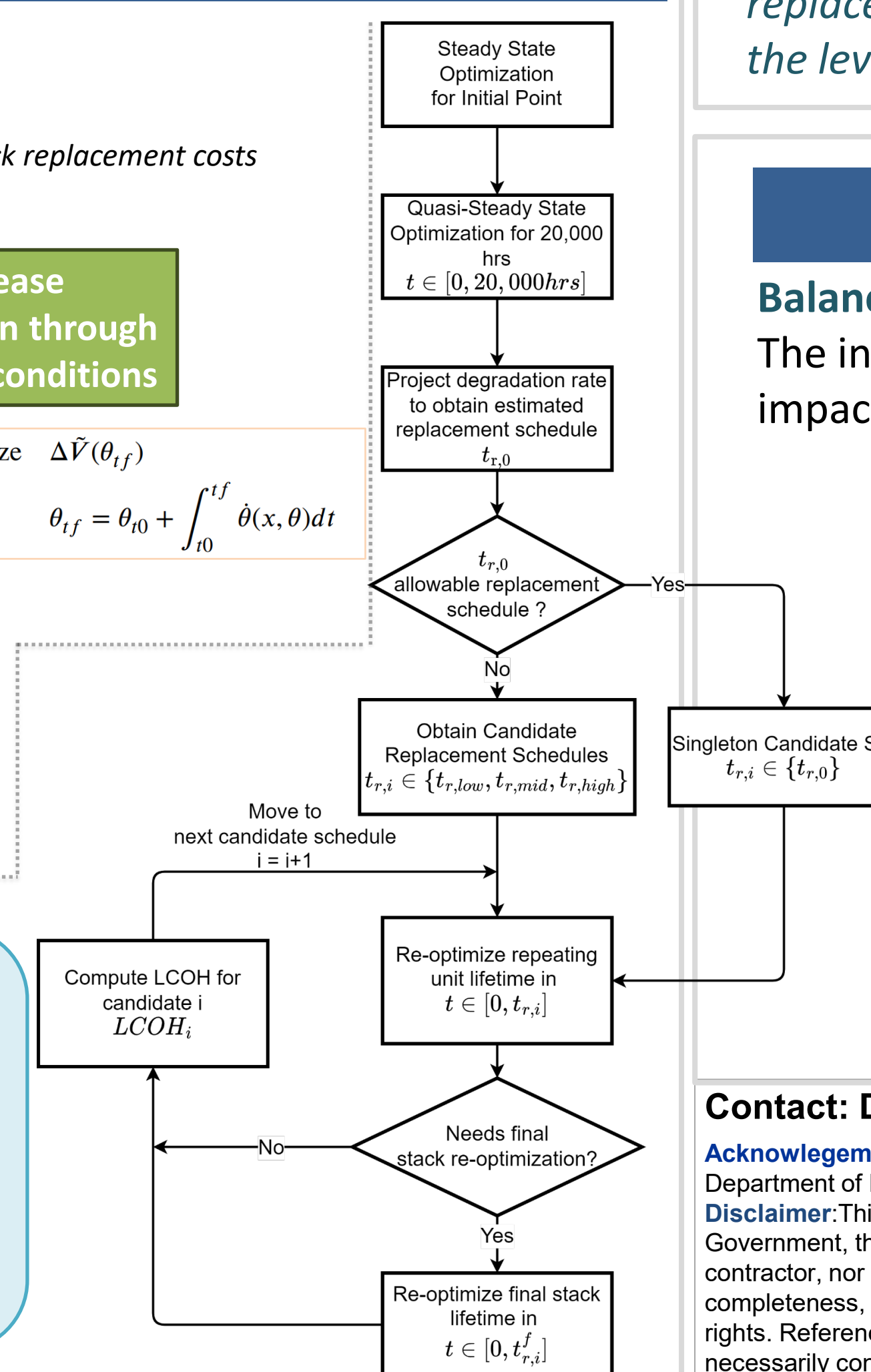
Optimization Formulation



minimize $LCOH$
where $LCOH = \frac{CRF (CC_{BOP} + \sum_{i=1}^R CC_{stack} F P_i) + OC + EC}{m_{H_2, lifetime}}$

Algorithm to determine optimal replacement schedule

- Re-optimize stack for different candidate replacement schedules, in parallel.
- Avoids binary decision variables.
- Stacks replaced when $\Delta \tilde{V}_{deg} > 50\%$.
- NLP subproblems solved using IPOPT.

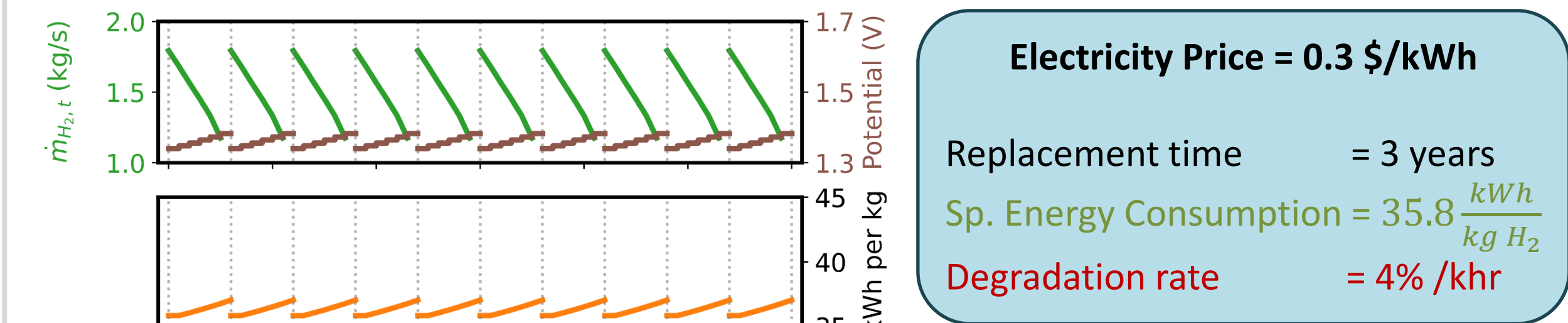
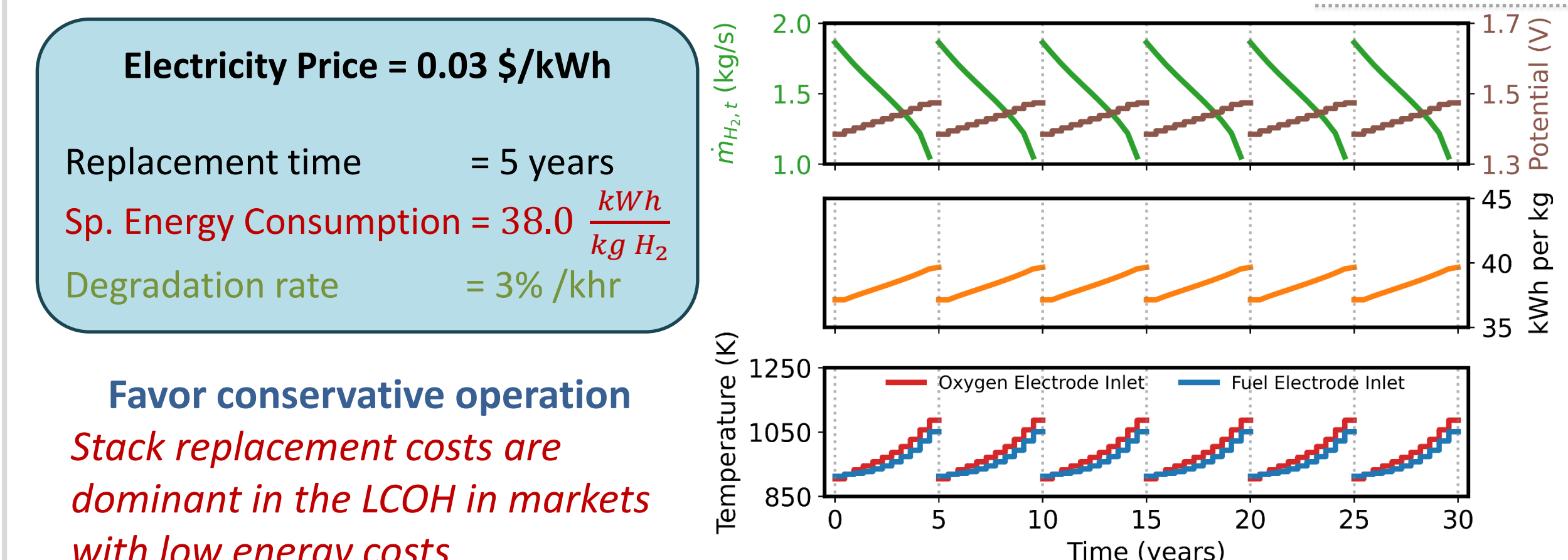


Selected Operating Schedules

Electricity Price = 0.03 \$/kWh

Replacement time = 5 years
Sp. Energy Consumption = 38.0 $\frac{kWh}{kg H_2}$
Degradation rate = 3% /khr

Favor conservative operation
Stack replacement costs are dominant in the LCOH in markets with low energy costs



Electricity Price = 0.3 \$/kWh

Replacement time = 3 years
Sp. Energy Consumption = 35.8 $\frac{kWh}{kg H_2}$
Degradation rate = 4% /khr

Favor aggressive operation
Energy costs are dominant in the LCOH in markets with high energy costs

Given electricity market information, this method automatically derives optimal replacement schedules and operating changes within a stack lifetime that minimize the levelized cost of H₂ (LCOH).

Takeaways

Balancing tradeoffs

The interplay of short-term efficiency and long-term degradation has a significant impact on the cost of H₂ production by an SOEC system.

Two distinct consequences of degradation

- Cell degradation impacts both system efficiency and cell thermals.
- Higher temperatures induced by degradation result in accelerated degradation rates and premature failure due to thermal stresses.

Insights from Optimization

- Combating degradation is both a materials design and an operational optimization problem.
- The tradeoff between efficiency and replacement frequency is captured through the LCOH.
- Flexible long-term operating mode can result in lower LCOH when compared to traditional galvanostatic and potentiostatic operation.

Contact: Debangsu Bhattacharyya, Debangsu.Bhattacharyya@mail.wvu.edu

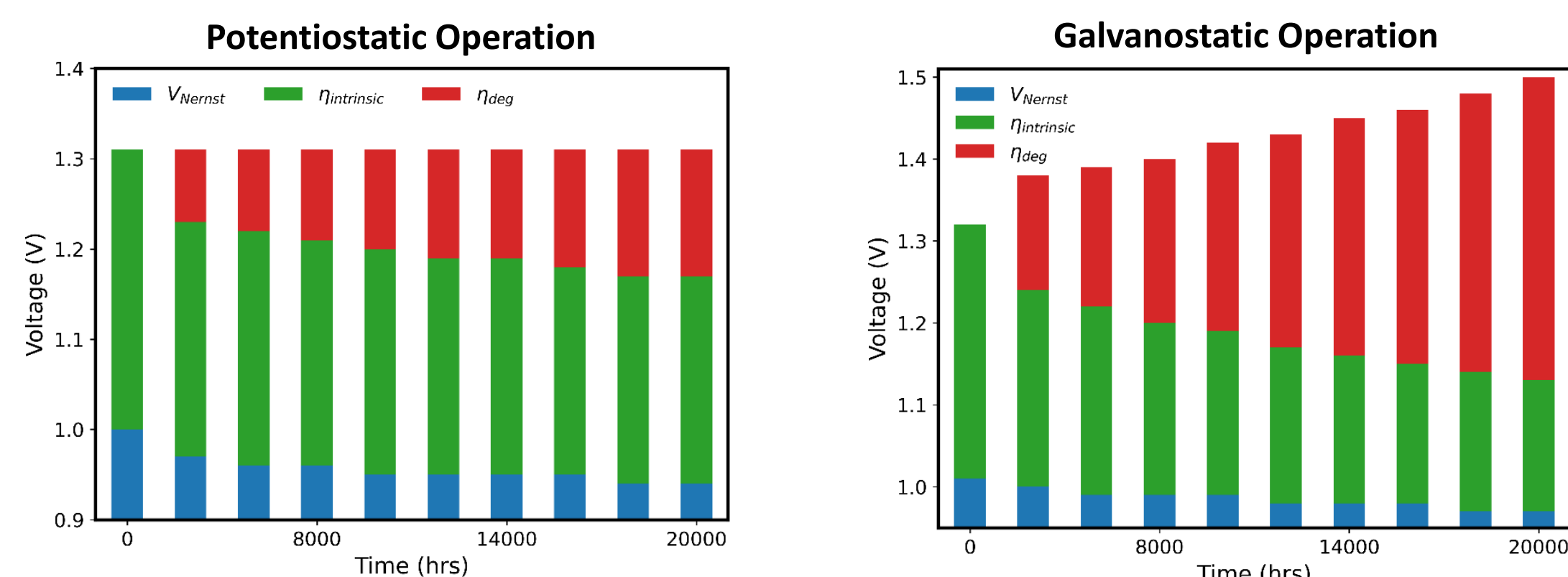
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Long-term Performance Degradation

Voltage losses increase the required voltage for electrolysis

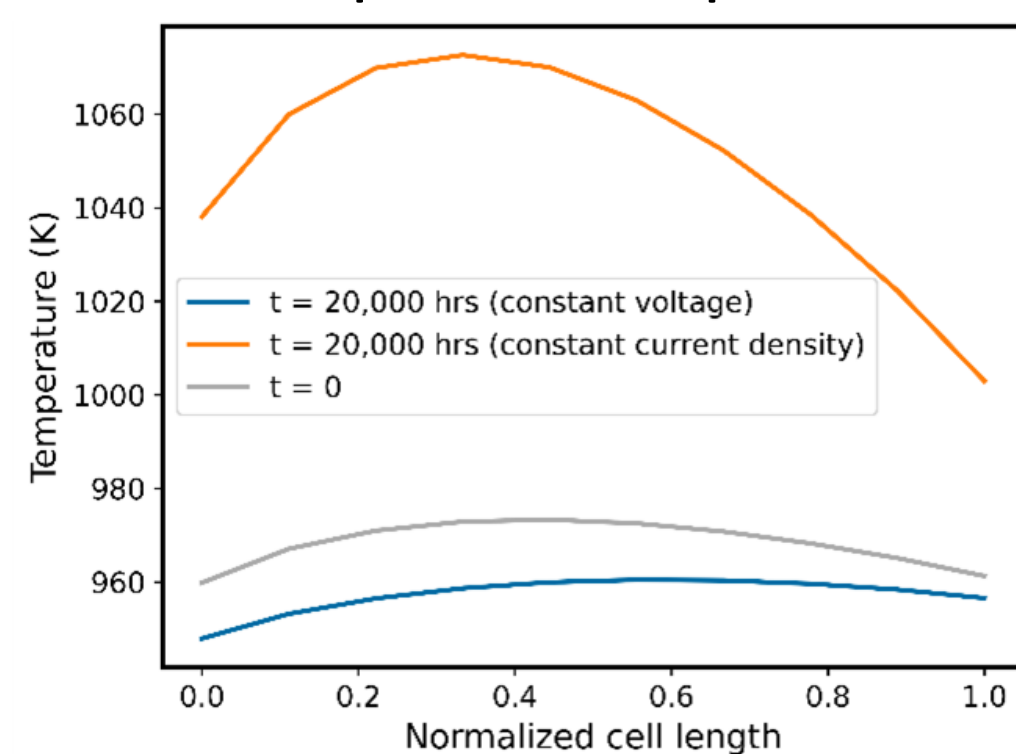
$$V_{cell} = \underbrace{V_{Nernst}}_{\text{Thermodynamic Minimum}} + \underbrace{\eta_{activation}}_{\text{Intrinsic Losses}} + \underbrace{\eta_{ohmic}}_{\text{Degradation Losses}} + \underbrace{\eta_{degradation}}_{\text{Degradation Losses}}$$

Distribution of degradation losses after 20,000 hours of degradation



Impact of degradation on cell performance

Impact on cell temperature



Impact of cell efficiency

